

Supporting Information.

A Quick Responsive Fluorogenic pH Probe for Ovarian Tumor Imaging

Ching-Hsuan Tung*, Jianjun Qi, Lingchuan Hu, Myung Shin Han, and Young Kim

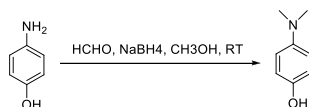
*Correspondence to:

Ching H. Tung, Weill Cornell Medical College, 413 East 69th Street, Box 290, New York, NY 10021. E-mail: cht2018@med.cornell.edu

1. Preparation of dye CypH-1
 - a. Synthesis of 4-(dimethylamino)phenol
 - b. Synthesis of CypH-1
 - c. ¹HNMR spectra of 4-(dimethylamino)phenol
 - d. Mass spectra of 4-(dimethylamino)phenol
 - e. ¹HNMR spectra of dye CypH-1
 - f. Mass spectra of dye CypH-1
 - g. ¹³CNMR spectra of dye CypH-1
2. Fig. S-1 Normalized absorption and emission spectra of dye CypH-1 in pH 4.0.
3. Fig. S-2 Quantum yield measurement of CypH-1.
4. Fig. S-3 LogP measurement of CypH-1

Preparation of dye CypH-1

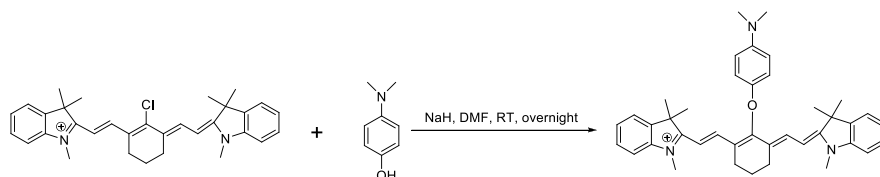
4-(dimethylamino)phenol



4-Aminophenol hydrochloride (1.33 g, 9.14 mmol) was dissolved in methanol/formaldehyde (30 ml/9 ml) at 0 °C. Sodium borohydride (3.50 g, 93 mmol) was added to the reaction solution slowly. The solution was stirred for one more hour, and then water (30 ml) was added. The reaction product was extracted with ethyl acetate (3 X), and washed with brine. The combined organic layers were dried over sodium sulfate. After concentration, the residue was purified by silica gel column, eluted with hexane and ethyl acetate (2/1) to give 780 mg of crystal solid, yield (62.4%). TLC: hexane/acetate=1/1, $R_f=0.4$. $^1\text{H-NMR}$ (300 MHz, MeOD): 6.80 (2H, d, $J=6.9\text{Hz}$), 6.71(2H, d, $J=6.9\text{Hz}$), 2.79 (6H, S). MS:138 ($\text{M}^+\text{+H}$).

Reference: JACS 2011, 133, 16970-6.

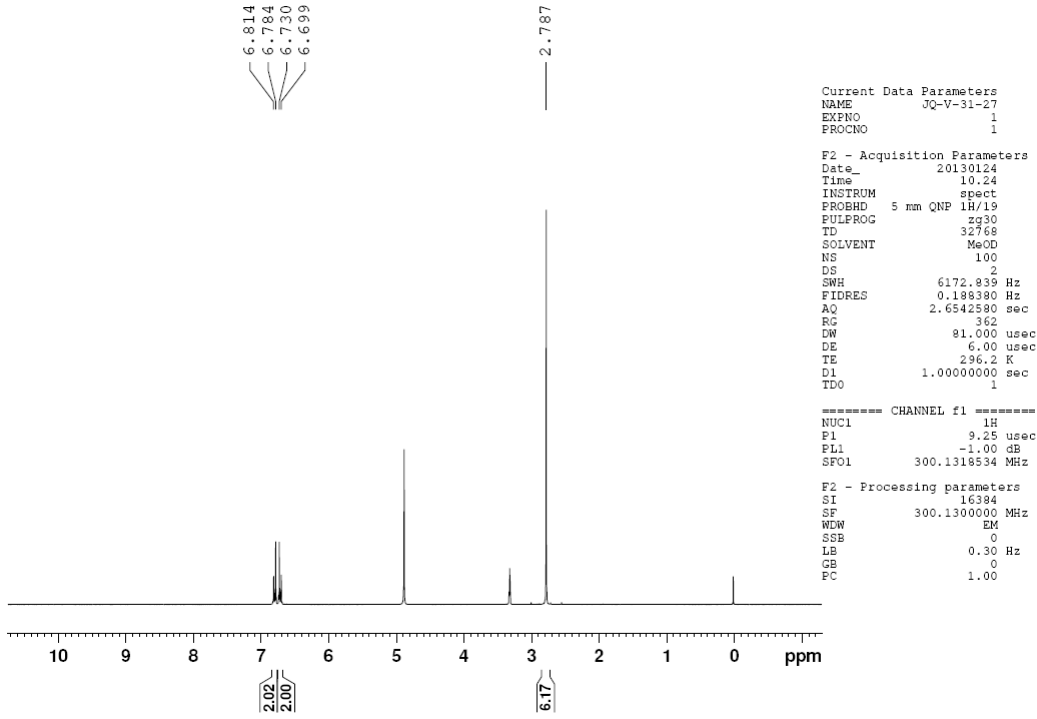
CypH-1



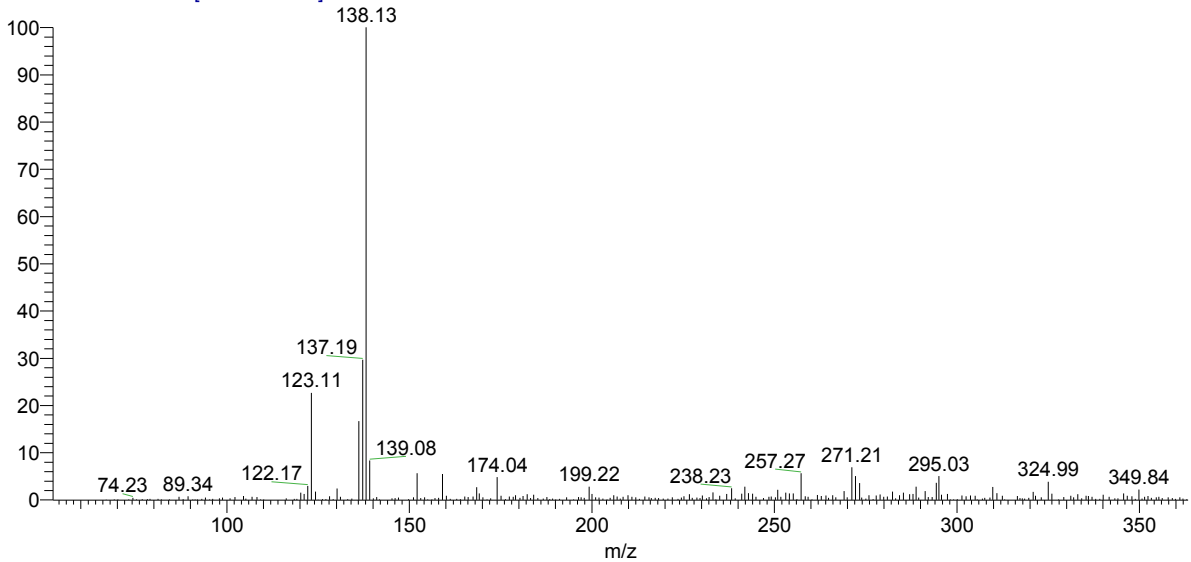
4-(Dimethylamino)phenol (57mg, 0.42mmol) in DMF (5 ml) was reacted with sodium hydride (17 mg) at RT for 15min. IR-775 (100 mg, 0.19 mmol, Aldrich, Dye content ~90%) was added slowly and stirred at RT overnight. The reaction was extracted with dichloromethane (DCM), and washed with brine. The product in DCM layer was purified by silica gel column using DCM/MeOH (MeOH 0-10%) as an elution solvent. The yield is 57 mg, 48 %. TLC: DCM/MeOH=10/1, $R_f=0.25$. $^1\text{H-NMR}$ (300 MHz, CDCl_3): 7.87 (2H, d, $J=14.1\text{ Hz}$), 7.71 (2H, d, $J=7.8\text{ Hz}$), 7.41-7.35 (2H, m), 7.31 (2H, d, $J=6.3\text{ Hz}$), 7.25 (2H, d, $J=7.5\text{Hz}$), 7.21-7.16 (2H, m), 7.09 (2H, d, $J=7.8\text{ Hz}$), 6.00 (2H, d, $J=14.1\text{ Hz}$), 3.58 (6H, s), 3.13 (6H, s), 2.75-2.65 (4H, m), 2.05-2.00 (2H,m), 1.36 (12H, s). $^{13}\text{C-NMR}$ (125MHz, MeOD): 172.77, 163.72, 158.83, 142.55, 141.99, 140.78, 139.21, 128.72, 125.44, 122.28, 122.08, 116.23, 110.26, 100.12, 49.02, 45.67, 31.22, 27.63, 24.18, 20.92. MS: 584 (M^+).

Reference: Bioconjugate Chem. 2011, 22, 2227-36.

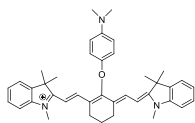
Spectra of 4-(dimethylamino)phenol



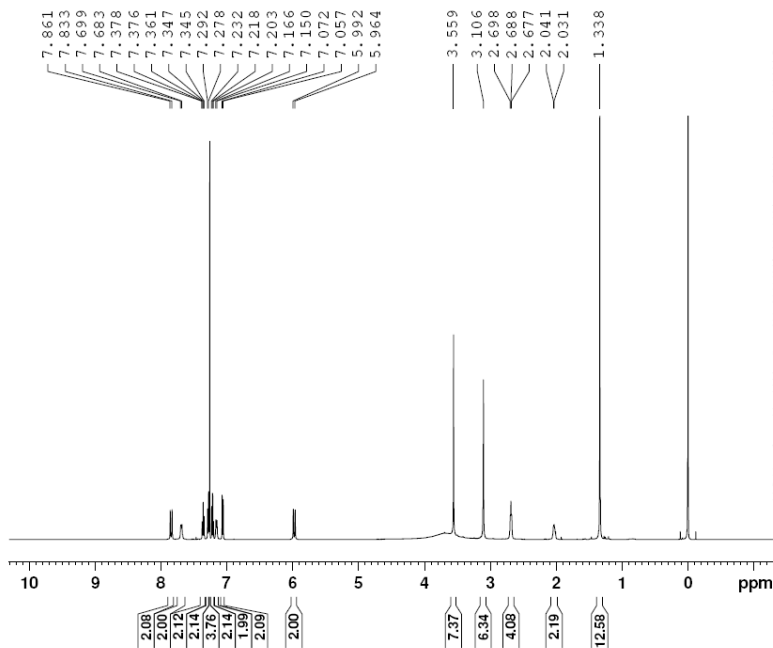
07122012-1_120310132051 #144 RT: 0.1 NL: 3.40E4
 T: ITMS + c ESI Full ms [50.00-800.00]



Spectra of CypH-1



PROTON CDC13 /opt/topspin ctung 20



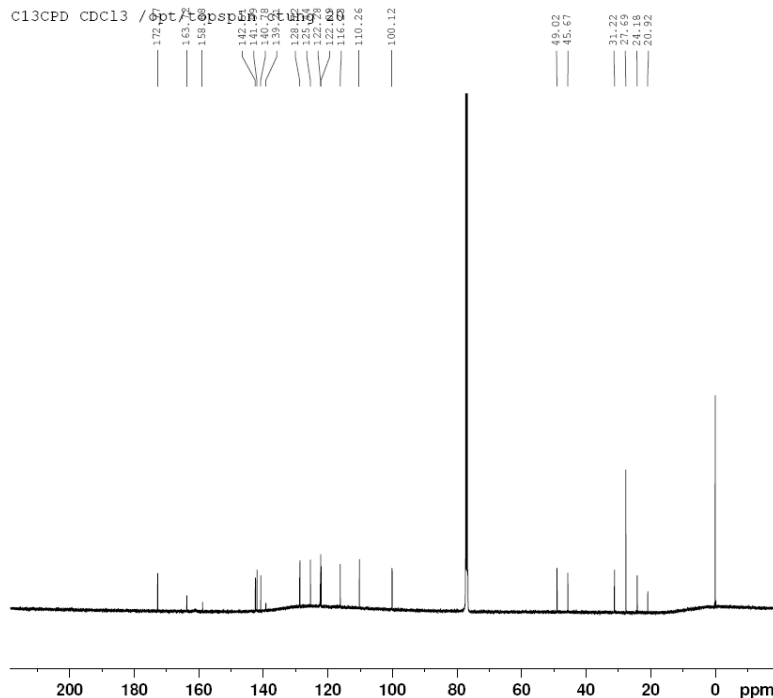
```
Current Data Parameters
NAME      JQ-V-31-27_500MHz
EXPNO    1
PROCNO   1

F2 - Acquisition Parameters
Date_    20130128
Time     17.15
INSTRUM  spect
PROBHD   5 mm CPPBBO BB
PULPROG  zg30
TD       65536
SOLVENT  CDCl3
NS       128
DS       4
SWH      10000.000 Hz
FIDRES   0.152588 Hz
AQ       3.2768500 sec
RG       67.4
DE       50.000 usec
TE       298.2 K
D1       1.00000000 sec
TDO      1

===== CHANNEL f1 =====
SFO1     500.1330885 MHz
NUC1     1H
P1       12.00 usec

F2 - Processing parameters
SI       65536
SF       500.1300124 MHz
WDW      EM
SSB      0
LB       0.30 Hz
GB       0
FC       1.00
```

C13CPD CDC13 /opt/topspin ctung 20



```
Current Data Parameters
NAME      JQ-V-31-27_500MHz
EXPNO    2
PROCNO   1

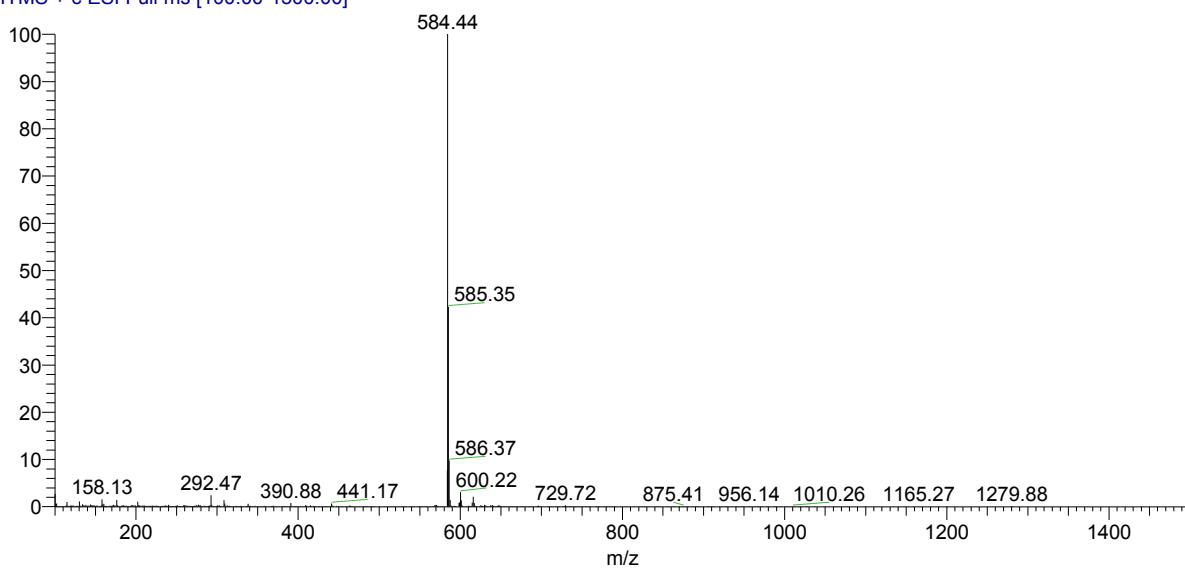
F2 - Acquisition Parameters
Date_    20130128
Time     22.32
INSTRUM  spect
PROBHD   5 mm CPPBBO BB
PULPROG  zg30
TD       65536
SOLVENT  CDCl3
NS       6000
DS       4
SWH      29761.904 Hz
FIDRES   0.454131 Hz
AQ       1.1010548 sec
RG       191.79
DE       16.800 usec
TE       298.2 K
D1       2.00000000 sec
D11     0.03000000 sec
TDO      1

===== CHANNEL f1 =====
SFO1     125.7703637 MHz
NUC1     13C
P1       10.00 usec

F2 - Processing parameters
SI       32768
SF       125.7577886 MHz
WDW      EM
SSB      0
LB       1.00 Hz
GB       0
FC       1.40
```

07132012-1_120310132051 #1 RT: 0.00
T: ITMS + c ESI Full ms [100.00-1500.00]

NL: 6.89E4



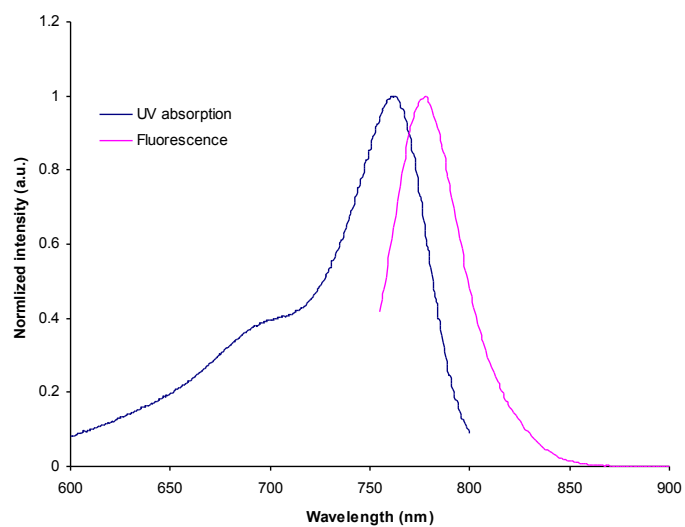


Fig S-1. Normalized absorption (blue) and emission (red) spectra of CypH-1 at pH=4 PBS buffer, $E_{x_{max}} = 760$ nm; $E_{m_{max}} = 777$ nm.

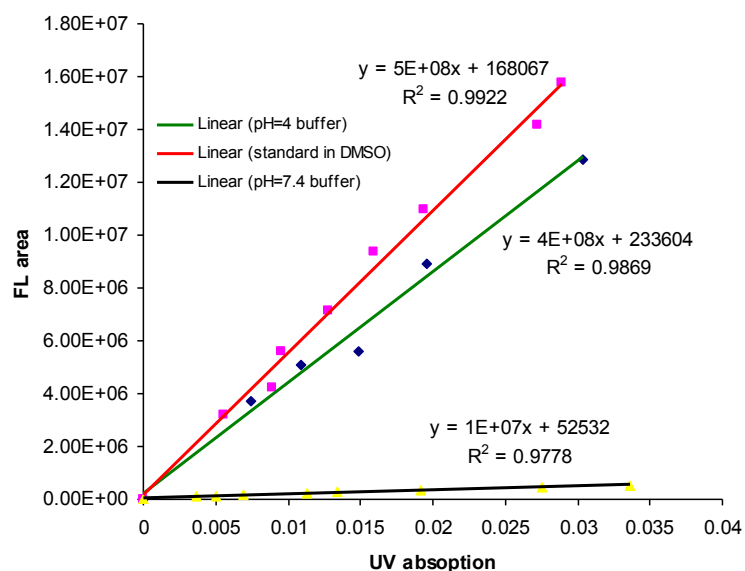


Fig S-2. Quantum yield measurement using indocyanine green in DMSO as a reference ($\Phi=0.106$) following a published protocol (Pure Appl. Chem., 2011, 83, 12, 2213–2228).
 $\Phi_{\text{pH}4.0} = \Phi_{\text{Standard}}[G_{\text{radx}}/G_{\text{rads}}][\eta_{\text{H}_2\text{O}}/\eta_{\text{DMSO}}]^2 = 0.106[4 \times 10^8/5 \times 10^8][1.33/1.479]^2 = 0.06857$
 $\Phi_{\text{pH}7.4} = \Phi_{\text{Standard}}[G_{\text{radx}}/G_{\text{rads}}][\eta_{\text{H}_2\text{O}}/\eta_{\text{DMSO}}]^2 = 0.106[0.1 \times 10^8/5 \times 10^8][1.33/1.479]^2 = 0.001714$

LogP measurement

Standard curve preparation. CypH-1 stock solution (200 μ g/ml) was prepared in MeOH/H₂O (1/3), then diluted to 100, 50, 10, 5 and 1 μ g/ml with pure water for a standard curve. The peak area at each concentration was determined using a reverse phase C18 column, (Phenomenex luna, 150X4.60 mm, 5 μ m) with mobile phase A: 0.1% TFA aqueous solution and mobile phase B: 0.1% TFA acetonitrile. The detector was set at 760 nm.

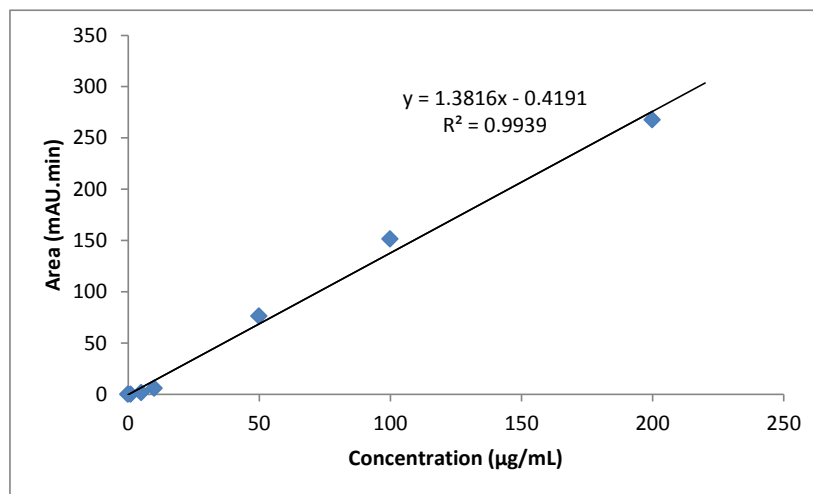


Fig S-3. The standard curve of CypH-1

		Peak area	Average Peak area (5 μ L)	Concentration (μ g/ml)
In octanol (injection 5 μ L)	1	159.7	156.03	113.24
	2	154.3		
	3	154.1		
In pH=7.4 PBS buffer (injection 40 μ L)	1	1.2	1.27/8 = 0.1588	0.42
	2	1.4		
	3	1.2		

$$Y = 1.3816X - 0.4191$$

$$C_{(\text{octanol})} = (156.03 + 0.4191) / 1.3816 = 113.24 \mu\text{g/ml}$$

$$C_{(\text{water})} = (0.1588 + 0.4191) / 1.3816 = 0.42 \mu\text{g/ml}$$

$$P = C_{(\text{octanol})} / C_{(\text{water})} = 113.24 / 0.42 = 269$$

$$\log P = \log [C_{(\text{octanol})} / C_{(\text{water})}] = 2.43$$